



Solution Q7

Chapter 22, Serway; **ABSOLUTELY NO CHEATING!**

Please write the answers on the blank space or on the back of this paper to save resources.

1.

(a) In an adiabatic process $P_f V_f^\gamma = P_i V_i^\gamma$. Also, $\left(\frac{P_f V_f}{T_f}\right)^\gamma = \left(\frac{P_i V_i}{T_i}\right)^\gamma$

Dividing the second equation by the first yields $T_f = T_i \left(\frac{P_f}{P_i}\right)^{(\gamma-1)/\gamma}$

Since $\gamma = \frac{5}{3}$ for Argon, $\frac{\gamma-1}{\gamma} = \frac{2}{5} = 0.400$ and we have

$$T_f = (1073 \text{ K}) \left(\frac{300 \times 10^3 \text{ Pa}}{1.50 \times 10^6 \text{ Pa}} \right)^{0.400} = \boxed{564 \text{ K}}$$

(b) $\Delta E_{\text{int}} = nC_V \Delta T = Q - W_{\text{eng}} = 0 - W_{\text{eng}}$, so $W_{\text{eng}} = -nC_V \Delta T$,

and the power output is

$$P = \frac{W_{\text{eng}}}{\Delta t} = \frac{-nC_V \Delta T}{\Delta t} \text{ or}$$

$$= \frac{(-80.0 \text{ kg})(1 \text{ mol} / 0.0399 \text{ kg})\left(\frac{3}{2}\right)(8.314 \text{ J/mol} \cdot \text{K})(564 - 1073) \text{ K}}{60.0 \text{ s}}$$

$$P = 2.12 \times 10^5 \text{ W} = \boxed{212 \text{ kW}}$$

(c) $e_c = 1 - \frac{T_c}{T_h} = 1 - \frac{564 \text{ K}}{1073 \text{ K}} = 0.475$ or $\boxed{47.5\%}$